FODOR, G., AND OTHERS

Determination of the absolute configuration of some tertiary amines and some quaternary ammonium salts. In German.

p. 62. (ACTA UNIVERSITATIS SZECEDIENSIS) Vol. 2, no. 1/4, 1956 Budapest, Hungary \$2.2900

SO: Monthly Index of East European Accessions (FEAI) LC, Vol. 7, No. 3, March 1958

FODOR, G.

A new reaction of the cyclization of amino alcohols; preparation of 20imido-4, 5-cyclopentano-1, 3-oxazolidine. In French.

p. 74. (ACTA UNIVERSITATIS SZEGEDIENSIS) Vol. 2, no. 1/4, 1956 Budapest, Hungary 52 5660

SO: Monthly Index of East European Accessions (EEAI) LC, Vol. 7, No. 3, March 1958

FORDIR, G. HUNGARY/Organic Chemistry - Natural Substances and Their

Synthetic Analogues.

G-3

Abs Jour

: Ref Zhur - Khimiya, No 7, 1958, 21601

Author

: G. Fodor, I. Sallay, F. Dutka

Inst

Title

: Quaternary Ammonium Salts Derived of (-)-Lupinine.

Orig Pub

: Acta phys. et chem. Szeged, 1956, 2, No 1 -4, 77-79

Abstract

: The configuration of the oxymethyl group with respect to the N atom in (-)-lupinine (I) was studied. Epimer iodides were prepared by the action of CH2ICOCC2H5 (II) on I.

The epimer (III), melting point 154° , $\sqrt{\alpha}$ 7^{24} D = -49.06° (c = 1.591), was prepared at about 20° of 1.07 g of I and 1.284 g of II in 3 mlit of absolute C6H6. The epimer (IV), melting point 148 to 150°,

1.05), was obtained of 0.354 g of I and 0.428 g of II in

Card 1/2

Univ Szeged, Hung.

HUNGARY/Organic Chemistry - Natural Substances and Their Synthetic Analogues.

G-3

Abs Jour

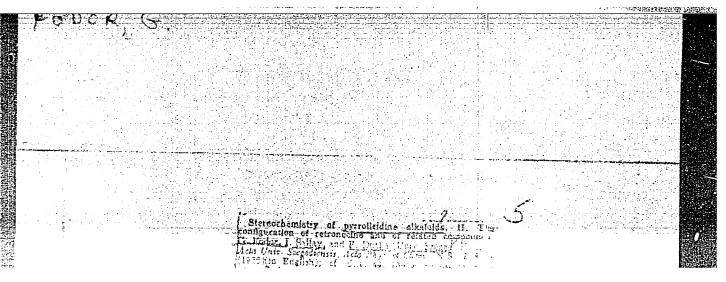
: Ref Zhur - Khimiya, No 7, 1958, 21601

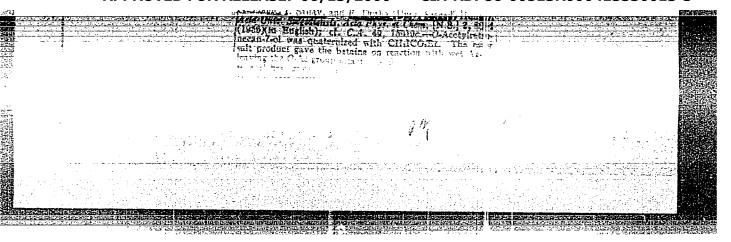
1 mlit of absolute alcohol and 3 mlit of absolute CoHo at 950 (22 hours in a sealed tube, after which 43 hours at about 200). III was transformed into a substance (V), mel-

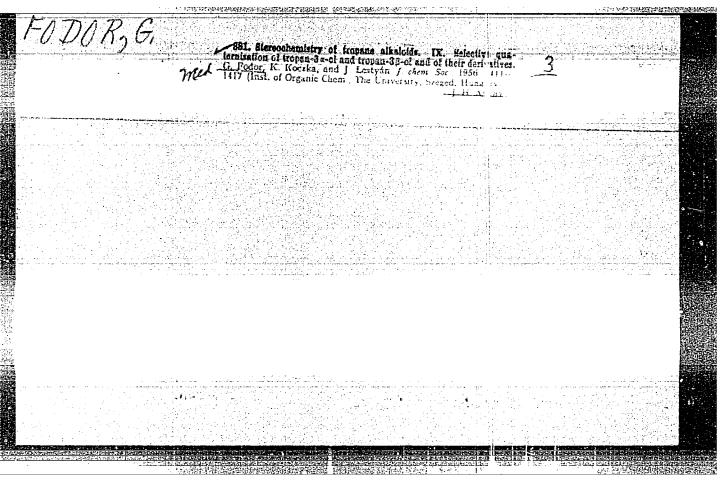
ting point 150 to 152°. $\left[\triangle \right]^{24}D = +16.67^{\circ}$ (c = 1.5), by boiling in 10 mlit of water in a sealed tube (24 hours at 95°). III produces betaine, melting point 244°,

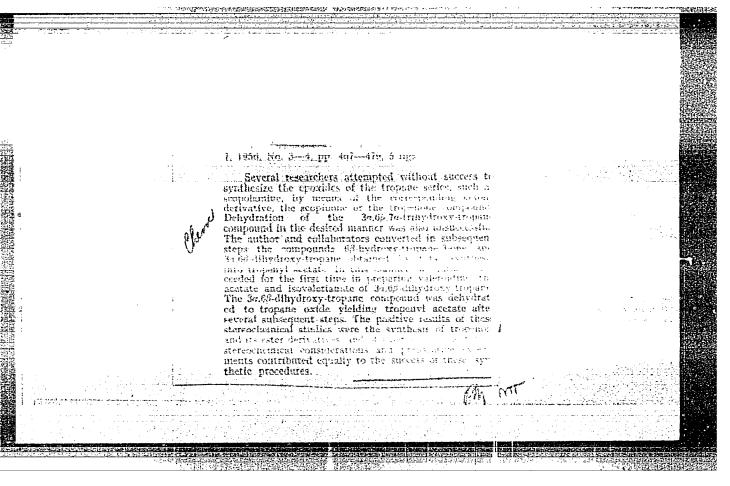
 $\int d^{26}D = +9.91^{\circ}$ (c = 1.029), by the action of an excessive amount of Ag_2O in 25 mlit of water (2 hours of shaking) and following boiling (5 hours). The boiling of betaines derived of III, IV and V with HBr or HI does not result in lactonization. All $\int d^{2}D$ -s were measured in water.

Card 2/2









FODOK, 打.

USSR / Organic Chemistry. Theoretical and General Problems of Organic Chemistry.

I-I

Abs Jour : Ref Zhur - Khimiya, No 6, 1957, No 18994

Author : Fodor G., Kochka K., Leshtian I., Tot I., Khal'mosh G.,

Kovach O., Vinche V.

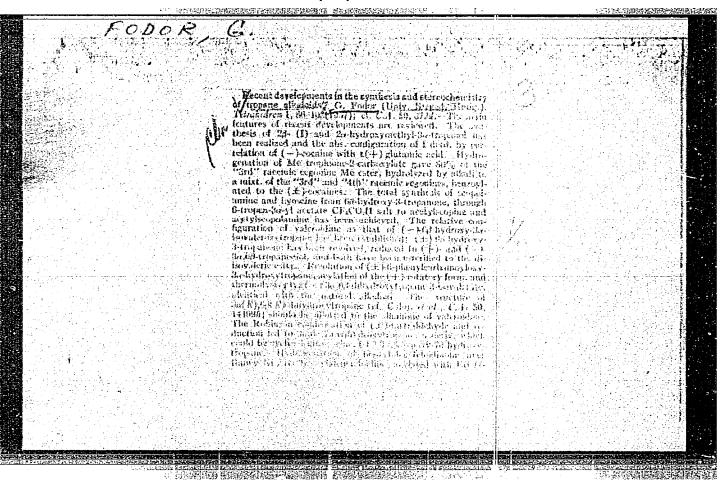
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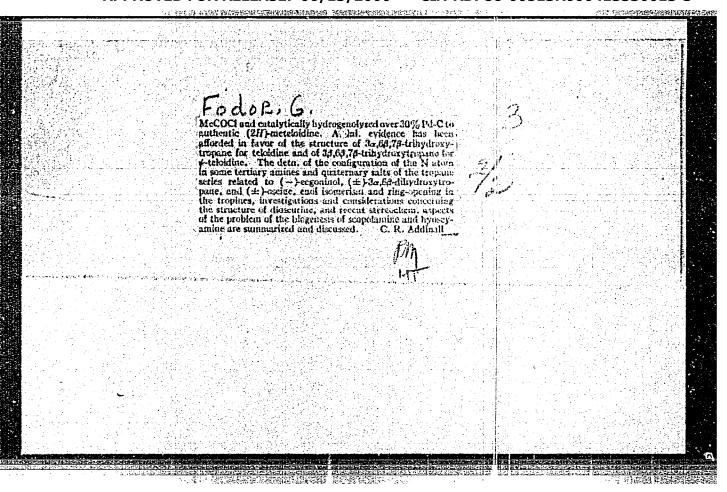
Title : *bsolute Configuration of Scme Tertiary Amines and Tetraamonium Salts.

Orig Pub : Uspekhi khimiyi, 1956, 25, No 7, 894-902

Abstract: Review of the work by the authors on the study of the spherical orientation of the bonds of nitrogen and the determination of absolute and relative configuration of tertiary amines and salts of tetraamonium bases in Bitliography with 24 titles.

Card : 1/1





HUNGARY/Organic Chemistry. Theoretical and General Questions on Organic Chemistry.

G-l

Abs Jour: Ref Zhur-Khim., No 13, 1958, 43177.

Author : Fodor Gabor, Kovacs Odon, Toth Jozsef, Koczka

Karoly, Koczor Istvan, Vincze Iren W., Lestyan

Janos, Halmos Miklos, Dobo Pal.

Inst

Title : Recent Methods and Advances in Stereochemistry of

Organic Compounds.

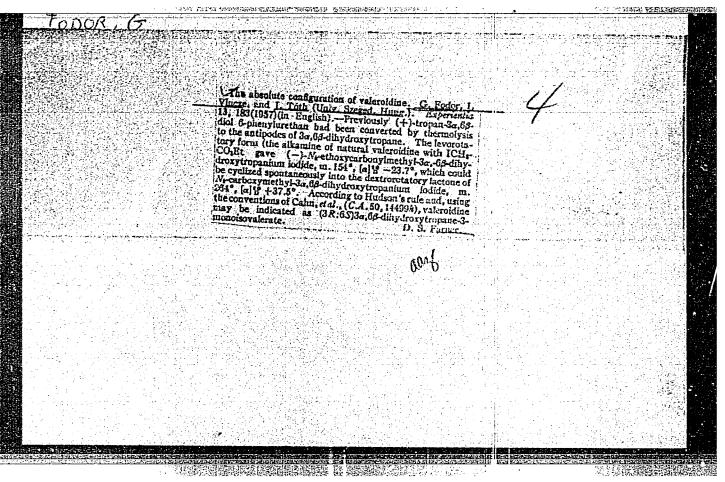
Orig Pub: Magyar tud. akad. Ken. tud. oszt. kozl., 1957, 9,

No 1, 77-91.

Abstract: A review, mostly of the work of the athors. Diblio-

graphy 58 references.

Card : 1/1



ACTA CHIMICA

Academiae Scientiarum Hungari ae

Vol. 13, Nrs. 1-2, 1957

THE SPEREOCHEMICAL COURSE OF THE CONVERSION

OF 2-UREIDO ALCOHOLS INTO ONAZA LIDINES, H.

REARRANGEMENT OF N'THIOUREIDO / LCOHOLS

K, KOCZKA and G: FODOR

(Institute of Organic Crimitics, University of Surger)

Received April 1, 1856

SUMMARY

(The stereochemical course of the convenient of N-thiocarpoint (2-timino sleukola particularly that of Organical Institute of the Course of the Cou

Survey of the Control	
ACTA CHIMICA	
Acadomiae Scientarium Hung ricae Vol 13. Wrs 1-2, 195	uE30
SYNTHETIC CONFIRMATION OF THE MECHANISM OF N=O ACYL MIGRA YONS	
PREPARATION AND REARRANGEMENT OF THE INTERMEDIATE 2,5-DIPHENYL-3,4-DIMETHYL-µ-HYDROXY-1,3 OXAZOLIDINE K, KOCKA and G. Fodor Granitate of Organic Chemistry, University of Secreti Received April 3, 1956 Granitative of the method ranged by affect on the other position of the method ranged by affect on the other position of protyphromic and. Namery, this decomposition of the control of the description of perceptations mentioned to State 11.	he //

YUGOSIAVIA/Organic Chemistry. deperal and Theoretical
Topics of Organic Chemistry.

G

Abs Jour: Ref Zhur-Khimiya, No 22, 1958, 73924.

Author : Gabor Fodor, Eva Fodor-Vraga, Arpad Furka.

Title : A Kinetic Contribution to the Knowledge of

Orig Pub: Croat. chem. acta, 1957, 29, No 3-4, 303-312.

Abstract: With a view to investigate the influence of spatial factors on the mechanism of N -> 0 transposition of the acyl group in N-substituted CV -amine alcohols, the rearrangement of cis- and trans-2-benzamidocyclo-hexanols-1 (I and II) and cis-2-benzamidocyclopenta-nol-1 (III) into cis- and trans-2-benzoyloxycyclohexylam

amines and cis-2-benzoyloxycyclopenylamine corres-

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Card : 1/5

YUGOSLAVIA/Organic Chemistry. General and Theoretical Topics of Organic Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 22, 1958, 73924.

pondingly under the action of HCl in dioxane was studied. The rate of the reaction with III was measured at 12 to 42° by the determination of the free amine, and that of the reactions with I and II were measured at 71 to 91° by the alkalimetric titration of the excess of HCl as well. Comparing the data for I, II and III after extrapolating them to 25° with the bibliographical values of the reaction rates of N-benzoyephedrine, cis- and trans-2-acetamidocyclohexanols-1 and cis- and trans-2-N-acetylinozamins (IV), the authors arrive at the conclusion that the transposition rate is determined mainly by the structure of the carbon framework of the alcohol, but not by the character

Card : 2/5

YUGOSLAVIA/Organic Chemistry. General and Theoretical.
Topics of Organic Chemistry:

G

Abs Jour: Ref Zhur-Khimiya, No 22, 1958, 73924.

of the solvent or of the migrating group, the rate ratio of the arylalifatic, cyclopentanic, cyclo-hexanic and isoaminic derivatives being 1000: 1000: 20: 1 correspondingly. The lesser reaction rates of I and II as compared with III is explained in accordance with the magnitudes of thermodynamic potential changes (I = 24.0, II = 24.3, III = 20.2 kcal per mole) by a lesser probability of intramolecular collisions in the cases of I and II in consequence of the existing conformation equilibrium. The cis-forms are 4 to 6 times more reaction capable than the trans-forms, because the latter can regroup only at the di-E arrangement of the amino and oxy groups, while the E,A, as well as the A,E conformations react in the cis-forms.

Card : 3/5

YUGOSIAVIA/Organic Chemistry. General and Theoretical Topics of Organic Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 22, 1958, 73924.

The above is confirmed by the difference between the activation energy values \triangle E (I - 15.02, II - 17.21 kcal per mole). The value of \triangle E of III (12.89 kcal) corresponds seemingly only to the transposition energy of the aci-group, and the increase of \triangle E of I and II is caused by the energy of the conformation conversion. The proposed mechanism of the regroupment with configuration preservation consists in an electrophilic attack by the proton of the carbonyl 0 and a following nucleophilic attack by the hydroxyl 0 of the carbonyl C with the formation of an intermediary cyclic complex. In accordance with the above, the little reaction capacity of IV can be explained by the difficulty of a nucleophilic attack

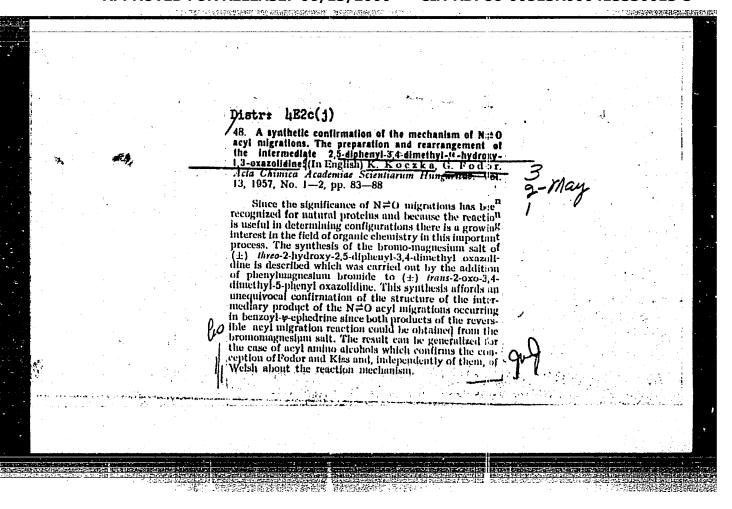
Card : 4/5

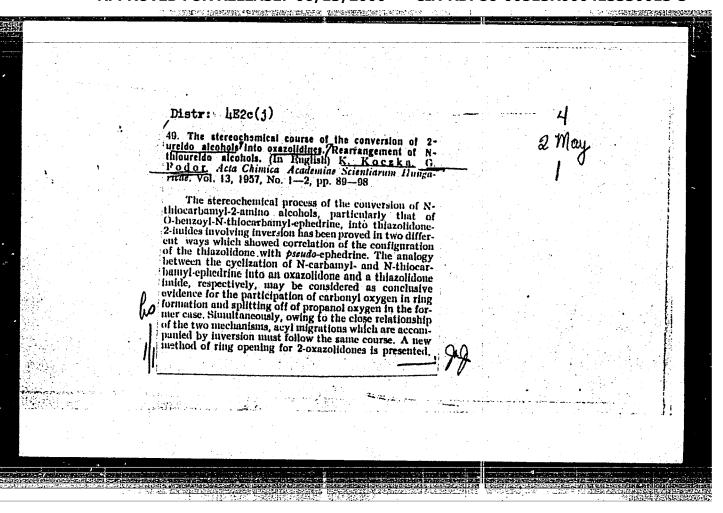
YUGOSIAVIA/Organic Chemistry. General and Theoretical Topics G of Organic Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 22, 1958, 73924.

in consequence of the participation of the hydroxyl 0 in the formation of the hydrogen bond. The reaction of the corresponding III trans- derivative does not agree with that mechanism and could not be studied, because it proceeds with a Walden inversion.

Card : 5/5





HUNGARY/Analytical Chemistry. Analysis of Inorganic

Compounds.

Abs Jour: Ref Zhur-Khimiya, No 21, 1958, 70551.

Author : Fodor.

Inst : Akad. Kem.

Title : Determination of Uranium by a Combined Method of

Ion Exchange and Complexometry.

Orig Pub: Magyar tud. akad. Kem. tud. Oszt. Kozl., 1958,

9, No 4, 463-470.

Abstract: No abstract.

Card : 1/1

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000413330013-3

HUNGARY/Chemical Technology. Chemical Products and Their Applications. Industrial Organic Synthesis.

H

Abs Jour: Ref Zhur-Khim., No 8, 1959, 28446.

Author: Fodor, G. and Beregi, L., and Kallay, F.

Inst : Hungarian Academy of Sciences.

Title : Results from Investigations on the Chemistry of

Furan in Hungary.

Orig Pub: Acta Chim Acad Sci Hung, 15, No 3, 315-323 (1958)

(in French with English and Russian surmaries)

Abstract: A survey of work done (1955) on the utilization of

furfurol (I) as an intermediate in the production of plastics and pharmceuticals. The following processes have been developed through the pilotplant stage: (a) the production of pyromicic acid

Card : 1/3

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HUNGIRY/Chemical Technology. Chemical Products and Their Applications. H Industrial Organic Synthesis.

Abs Jour: Ref Zhur-Khim., No 8, 1959, 28446.

(II) in yields of 96-98, by the oxidation of I with 02 or with air in the presence of Ca(OH)2 and using Ag20 as the catalyst (C); (b) the decarboxylation of II in the vapor phase (C: quinoline) with the separation of furan (III) from CO₂ by adsorption on charcoal; (c) the oxidation-decarboxylation of I in the vapor phase (C: oxides of heavy netals, particularly Pb); (d) the cleavage of the ring of derivatives of III, particularly of III itself, with H₂O₂ and HCl or with H₂SO₄, leading to the formation of malcic acid (IV) in the first case and of IV and succinic acid, in the second case. Industrial methods for the continuous hydro-

Card : 2/3

HUNGARY/Chemical Technology. Chemical Products and Their Applications. H Industrial Organic Synthesis.

Abs Jour: Ref Zhur-Khim., No 8, 1959, 28446.

genation of I to furfurol alcohol (C: Cu chromite) and of III to tetrahydrofuran have also been developed. -- Ya. Kantor.

Card : 3/3

207

FODOR, G.

SCIENCE

PERIODICALS. ACTA ZOOLOGICA. Vol. 64, No. 7/8 July/Aug. 1958 PAGYAR KEMIAI FOLYOTRAT

Fodor, G. Some newer applications of conformation analysis in the chemistry of hydrocarbons. p. 298

Monthly list of mast European Accessions (EEAI) LC, Vol. 8, No. 2, February 1959, Unclass.

FODOR, G.

SCIENCE

PERIODICAL: MAGYAR KEMIAI FOLYOIRAT. Vol. 64, no. 7/8, July/Aug. 1958

Fodor, G. Sterochemistry of Prins reaction and its application to ketomes. p. 301.

Monthly list of East European Accessions (EEAI) LC, Vol. 8, No. 2, February 1959, Unclass.

Fodor, G.		
	Constitution of trimethylsulfoxonium iodide. D. Banfi, G. Fodor, and L. Oivos (Hung. Acad. Sci., Budapest). Chem. & Ind. (London) 1959, 1102.—Chili with Me,SO	ME3T ME3T
	gave (Ci ⁴ H ₆ Me ₅ SO)] (I), which was transmethylated with C ₁ H ₈ N and quinoline by the procedure of Kuhn and Trischmann (C.A. 52, 14523g) for the radioinactive compd. The quaternary salts formed were found to be '/rd as active as I, which was evidence for sym. bonding of the 3 Me groups and hence for the S-oxo-S-trimethylsulionium salt constitution of the adduct. The results agreed with the structure suggested by K. and T. (loc. cit.) and by Smith and Winstein (C.A. 53, 4180e).	
	Card 1/1 aht	

KONDI, V.; IACOBESCU, A.; HAIAN, St.; FODOR, G.; MITRICA, Natalia.

An anticoagulant inhibiting thromboplastin formation.

Rumanian M, Hav. 4 no. 1:37-39 Ja-Mr '60.

(THROMBOPIASTIN)

(ANTICOAGULANTS pharmacol.)

BECK, Mihaly; BITE, Pal; BRUCKNER, Gyozo; CSENTES, Joznef; CSUROS, Zoltan; DEAK, Gyula; ERDEY-GRUZ, Tibor; ERDEY, Laszlo; FABIAN, Pal; FINALY, Istvan; FODOR, Gabor; FODORNE CSANYI, Piroska; GYORBIRO, Karoly; INZELT, Istvan; KUCSMAN Arpad; NEUMANN, Erno; PUNGOR, Erno; SCHNEER, Anna; SCHULEK, Elemer; SZABADVARY, Ferenc

Rules for the Hungarian chemical nomenclature and orthography. Kem tud kozl MTA 17 no.1/4:1-292 '62.

1. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja (for Hruckner, Csuros, Laszlo Erdey, G.Fodor, and Schulek). 2. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkesztoje (for Erdey-Gruz). 3. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" technikai szerkesztoje (for Finaly). 4. Muvelodesugyi Miniszterium (for Csentes). 5. Magyar Tudomanyos Akademia Helyesitasi Bizottsage (for Fabian). 6. Nehezipari Miniszterium (for Neumann).

FODOR, Gabor, akademikus

An account of my study trip to England and the German Federal Republic. Kem tud kosl MTA 18 no.2:325-335 '62.

1. Magyar Tudomanyos Akademia Sztereckemiai Kutato Ugoportja, Budapest, es "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja.

FODOR, Gabor, akademikus

An account of the symposium arranged on the 75th anniversary of the Belgian Chemical Society. Kem tud kozl MTA 18 no.4:605-609 162.

1. Magyar Tudomanyos Akademia Sztereokemiai Kutato Choportja, Budapest, es "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja.

Modern trend of organic chemical research and the significance of exploring reaction mechanism; an introduction to a series of articles. Magy kem lap 18 no.1:12-17 Ja '63. 1. Magyar Tudomanyos Akademia Sztereokemiai Kutatolaboratorium.

FODOR, Gabor Intramolecular rearrangements. Pt. 1. Magy kem lap 18 no.9: 414-422 S 163. 1. Magyar Tudomanyos Akademia Sztereokemiai Kutatolaboratorium; "Magyar Kemikusok Lapja" rovatvezetoje.

FODOR, Gabor Intramolecular rearrangements. Pt.2. Magy kem lap 18 no.12:593-597 D '63. 1. Magyar Tudomanyos Akademiai Kutato Intezet.

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FODOR, Gabor, akademikus; EEKE, Denesne; BITE, Pal, kandidatus; DOBO, Pal;
FARKAS, Lorant, kandidatus; F. VARGA, Eva; IEMPERT, Karoly, kandidatus;
OTVOS, Laszlo, kandidatus; SZANTAY, Csaba, kandidatus; URESCH, Ferenc

An account of the Prague Symposium on Natural Organic Compounds. Kem tud kozl MTA 19 no.1:95-103 '63.

1. Magyar Tudomanyos Akademia Sztereokemiai Kutato Csoportja, Budapest (for Fodor, Beke, Lempert, Otvos, Uresch). 2. Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalya (for Bite, Dobo, Farkas, F. Varga, Szantay). 3. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja (for Fodor).

FODOR, Gabor, akademikus; MORACSI, Tivadar; TOMASZ, Jeno

Present state of the chemistry of nucleotides. Kem tud kozl MTA 19 no.2:163-179 '63.

1. Magyar Tudomanyos Akademia Sztereokemiai Kutato Csoportja, Budapest. 2. "A Magyar Tudományon Akademia Kemiai Tudomanyok Osztalyansk Koslemenyei" szerkeszto bizottsagi tagja (for Fodor).

FODOR, Gabor, akademikus

An account of the 1962 Scientific Congress of the Chemical Society in the German Democratic Republic. Kem tud kozl MTA 19 no.3:357-362 '63.

1. Magyar Tudomanyos Akademia Sztereokemiai Kutato Csoportja, Budapest; *A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei* szerkeszto bizottsagi tagja.

FODOR, Gabor, akademikus

An account of the First Prague Conference on the Chemistry and Biochemistry of Nucleic Acids, arranged by the scientific academies of the socialist countries. Kem tud kozl MTA 20 no.4:471-472 163.

1. Magyar Tudomanyos Akademia Satereokemiai Kutato Ceoportja, Budapest; A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei szerkeszto bizottsagi tagja.

FODOR, Gaborne An account of my study trip to the Soviet Union. Kem tud kozi MTA 19 no.2:239-249 '63. 1. Magyar Tudomanyos Akademia Sztereokemiai Kutato Csoportja, Budapest.

FODOR, Gabor, akademikus

Research in stereochemistry, synthesis and biogenesis of tropane alkaloids conducted since 1955. Pt. 1. Kem tud kozl 20 no.3:336-373 '63.

1. Magyar Tudomanyos Akademia Sztereokemiai Kutato Csoportja, Budapest; "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja.

FODOR, Gabor, akademikus

Research in the stereochemistry, synthesis and biogenesis of tropane alkaloids since 1955. Pt.2. Kem tud kesi MYA 20 no.4: 441-467 63.

1. Magyar Tudomanyos Akademia Setereskemiai Kutato Csoportja, Budapest; "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztanlyanak Közlemenyei" szerkeszto bizottsagi tagja.

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ERDEY-GRUZ, Tibor, akademikus; BRUCKNER, Gyozo, akademikus; VAPGHA, Larzlo; KORACH, Mor, akademikus; FREUND, Mihaly, akademikus; FODOR, Gabor, akademikus; GERECS, Arpad, akademikus; SCHAY, Geza, akademikus; BITE, Pal, kandidatus; BOGNAR, Rezso, akademikus; FARKAS, Lorand, kandidatus

An account of the work of the Section of Chemical Sciences, Hungarian Academy of Sciences. Kem tud kozl MTA 22 no.2:109-152 '64.

1. Secretary, Section of Chemical Sciences, Hungarian Academy of Sciences, and Editor, "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei", Budapest (for Erdey-Gruz). 2. Editorial board member, "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" (for Bruckner, Korach, Freund, Fodor, Gerecs, Schay and Bognar). 3. Corresponding member, Hungarian Academy of Sciences, and Editorial board member, "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" (for Vargha).

FODOR, Gabor, akademikus

An account of my study trip to Northern Europe. Kem tud kozl MTA
22 no.2:289-291 '64.

1. Research Group of Stereochemistry, Hungarian Academy of
Sciences, Budapest, and Editorial board member, "A Magyar Tudomanyos
Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei".

DYKHOVA, Z.I.; MATYUSHINA, N.A.; MOSKVINA, M.M.; PROKOFTYEVA, G.P.; KHARLAMOV, V.T.; CHIRKOV, Ye.F.; FODOR, G.; FILLIF, I.

[Radioactive isotopes and labeled compounds; a catalog]
Radioaktivnye isotopy i mechenye soedineniia; katalog.
Moskva, Atomizdat, 1964. 341 p. (MIRA 18:1)

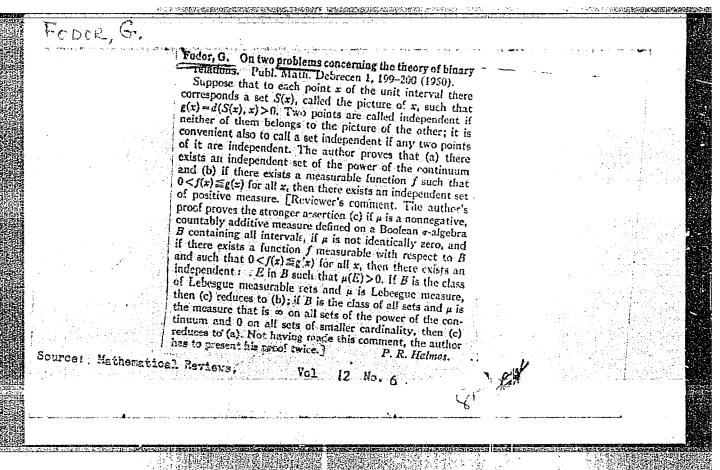
1. Sovet ekonomicheskoy vzaimopomoshchi. Postoyannaya kemissiya po ispol'zovaniyu energii v mirnykh tselyakh.

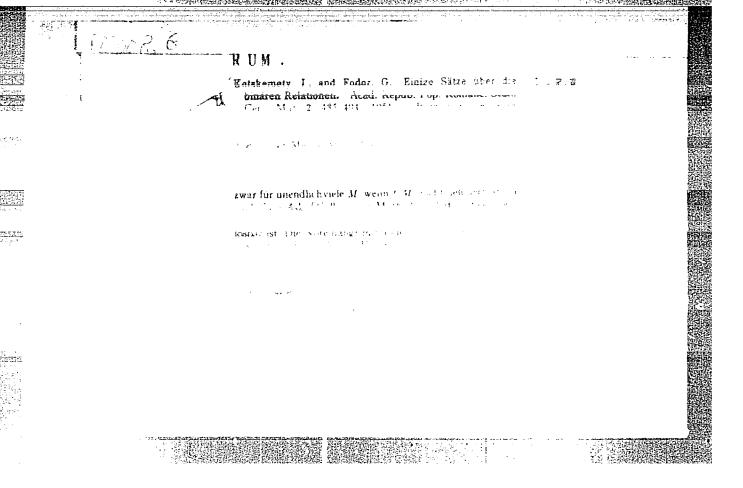
GALATEANU, I.; FODOR, G.; CHIOTAN, C.; CRISTU, M.

Obtaining 59Fe without a bearer. Studii cerc chim 13 no.10:643-652 0 '64.

1. Institute of Atomic Physics of the Rumanian Academy, Bucharest, P.O. Box 35.

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000413330013-3"





Fodor, G. Proof of a conjecture of P. Erdős, Acta Sci. Math. Szeged 14, 219-227 (1952). Soit E un ensemble non dénombrable de puissance m et n un nombre cardinal donné, tel que Ro≤n <m. Si RCE×E est une relation binaire entre éléments de E telle que card R(x) < n, R est coloriable avec moins de n couleurs (c'està-dire, il existe une relation d'équivalence $U \subset E \times E$ telle que Mathematical Review Ro UCA et card (E/U) ≤n). Ce théorème, qui avait été ! conjecture par P. Erdös [Proc. Amer. Math. Soc. 1, 127-141, June 1954 (1950), pp. 133-137; ces Rev. 12, 14] donne immédiatement Analysis une démonstration de la conjecture de Ruziewicz (c'est-àdire, il existe un sous ensemble $X \subset E$ tel que $R \cap (X \times X) \subset \Delta$ et card X=n) dans le cas où m ne peut être décomposé en une somme de n ou d'un nombre moindre de nombres cardinaux dont chacun est plus petit que m. J. Riguet.

Mathematical Reviews Vol. 15 No. 2 Fob. 1954 Analysis

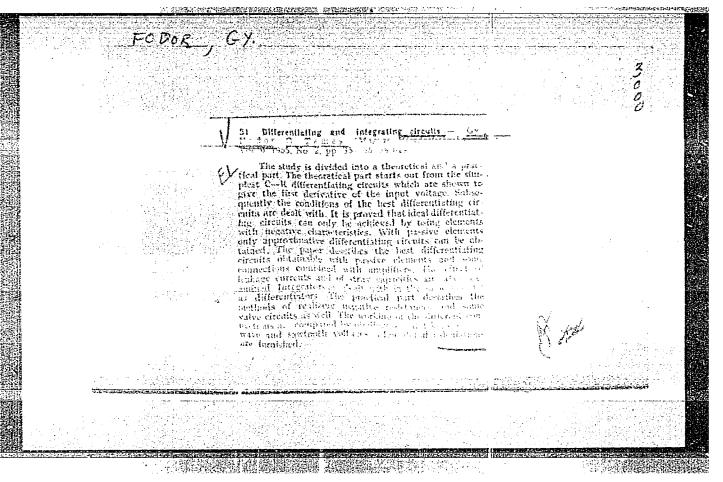
Mathematical Reviews Vol. 15 Ho. 2 For every ordinal number a, the following two propositions are equivalent: (1) $2^{N_0} = N_{a+1}$. (2) Let $|E| = 2^{N_0}$, and denote by B the class of all subsets of E of power 2. Then there exists a mapping, T, of B into E such that (a) if $F = |X_0| \ne B$, then either T(r) = x or T(r) = y, and (b) if $E \subseteq E$ and $|E_1| > N_0$, then E is equal to the union of the sects $r \in B$ for which $T(r) \in E$.

F. Bagemihl.

FODOR, G. - Koslemenyei - Vol. 5, no. 1, 1955.

Problem of the set theory. p. 57.

SO: Monthly list of East European Accessions, (EEAL), LC, Vol. 4, No. 9, Sept. 1955 Uncl.



FODOR, C.

FODOR, G. Generalization of a theorem of Alexandroff and Urysohn. In English. p. 204.

Vol. 16, No. 3/4, Dec. 1955. ACTA SCIENTIARUM MATHEMATICARUM SCIENCE Eudapest, Hungary

So: East European Accession, Vol. 5, No. 5, May 1956

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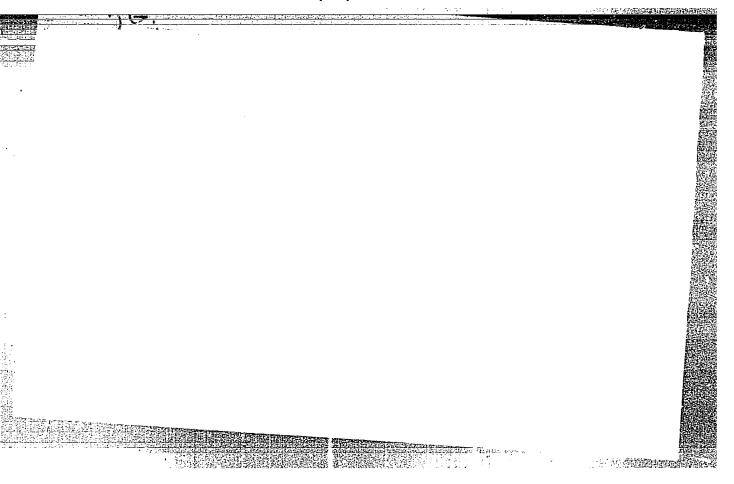
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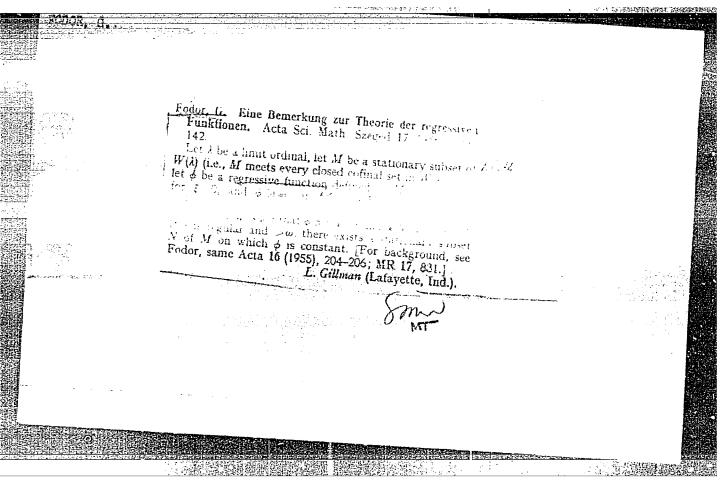
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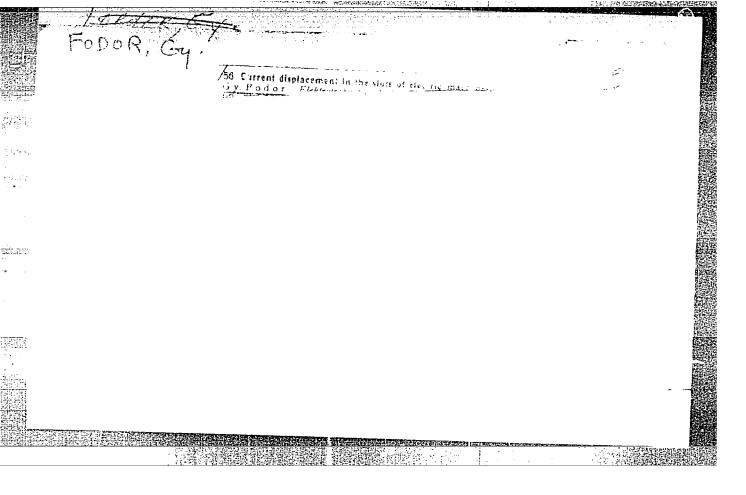
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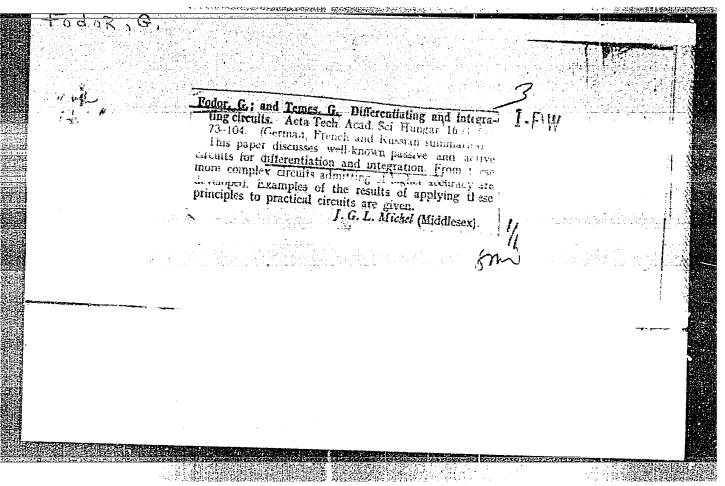
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Regulation of reactors. I. (To be contd.)

P. 582. (ENERGIA ES ATOMTECHNIKA.) (Budapest, Hungary) Vol. 10, No. 11/12, Nov./Dec. 1957

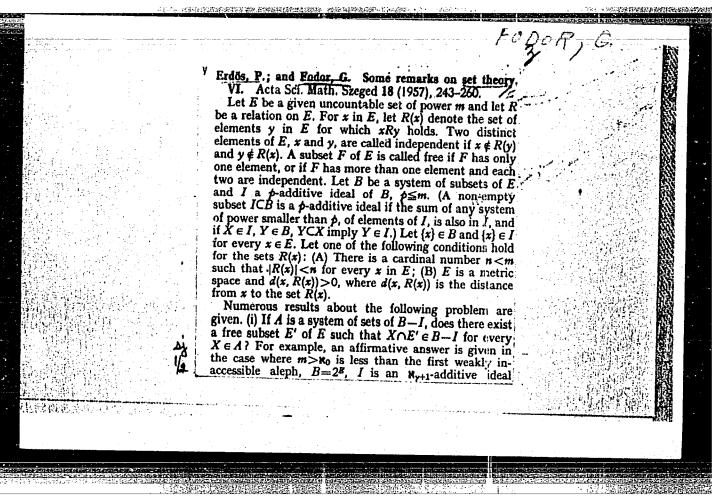
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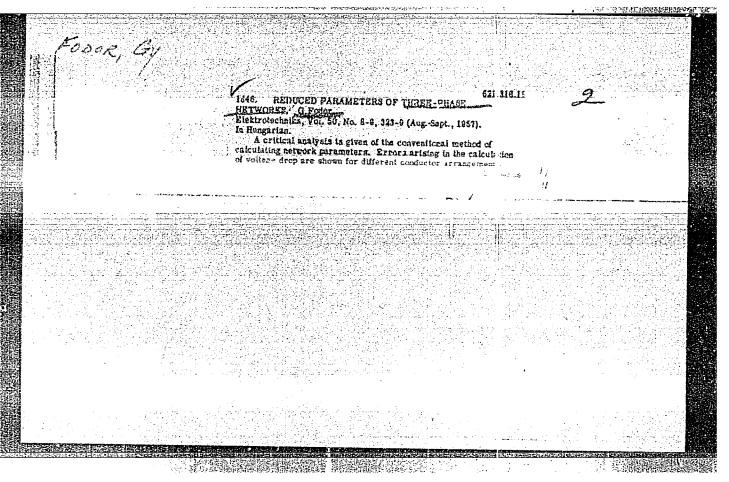


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HUNGARY/Nuclear Physics - Nuclear Power and Technology

C-8

Abs Jour : Ref Zhur - Fizike, No 4, 1959, No 7767

huthor : Fodor Gyorgy

Inst:

.... -

Title : Control of Reactors, Part II.

 $\mathtt{Ori}_{\mathcal{B}}$ Pub : Energia es atomtechn., 1958, 11, No 1-2, 1-8

Abstract: Survey article on the control of reactors. The following

problems are considered: self-regulation of the reactor, control rods, the reactor-control loop, the transfer function, programed regulation, and starting and stopping of

the reactor. -- V.I. Lend'yel

是34年的國際學院的第二個國際公司

Card : 1/1

18

HUNG:RY/Nuclear Physics - Nuclear Power and Technology

c-8

Abs Jour : Ref Zhur - Fizika, No 4, 1959, No 7736

Author : Fodor Gyorgy

Inst Title

: •

: Dictionary in Nuclear Engineering

作的學學問題的情情可能 多常學習

Orig Pub: Emergie es atomtechn., 1958, 11, No 1-2, 39-40

Abstract: An explanation is given of many terms pertaining to reactors.

Card : 1/1

16

HUNGARY/Nuclear Physics - General

C-1

Abs Jour: Ref Zhur - Fizika, No 3, 1959, No 4933

Author

: Fodor Gyorgy

Inst Title

: Systems of Units in Atomic Engineering

Orig Pub: Energia es Atomtechnika, 1958, 11, No 3, 138-142

Abstract: The author analyzes the MKS, CGS and the practical atomic system, the technical system of absolute units, and the so-called modified absolute system. A table of conversion of various quantities from one system to another is given. The author draws the following conclusion from his premises: for physical problems it is best to use the system MnJ, CGS, the practical atomic, and possible also the modified atomic; for technical problems it is best to use the MKS or the

technical system of units. -- V.I. Lend'yel

Card : 1/1

HUNGARY/Nuclear Physics - Nuclear Technology and Power Engineering C-8

Abs Jour: Ref Zhur - Fizika, No 4, 1959, No 5273

Author : Fodor Gyorgy

Inst :

Title : Dictionary of Nuclear Engineering [sic!]

Orig Pub: Energia es Atomtochnika, 1958, 11, No 3, 151-152

Abstract: Expressions are given for the critical demensions of the

sphere, cylinder, and tube as functions of the reactor para-

meter. Tables are given for the physical constants of

heavy water and of its chemical properties.

Card : 1/1

c-6

FODOR G.

HUNGARY/Nuclear Physics - Penetration of Charged and Neutral

Particles Through Matter

Abs Jour : Ref Zhur- Fizika, No 5, 1959, No 10186

Author : Fodor Gyorgy

Inst : Title : M

: Measurement of the Diffusion Length in Dedies Having a

Shape of a Prism, Cylinder, or Sphere

Orig Pub : Energia es Atomtechn., 1958, 11, No 4-5, 294-302

Abstract : No abstract

Card : 1/1

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FORCE, GY.

Temperature factor. p.650

EMERGIA ES ATOMTECHNIKA. (Energiagazdalkodasi Tudomenyos Egyesulet) Budapest, Hungary Vol. 11, no.9/10, Sept./Oct. 1958

Monthly List of East European Accessions (SEAI) IC., Vol. 8, no.7, July 1959 Uncl.

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Nuclear technical encyclopedia. p673.

ENERGIA ES ATOMTECHNIKA. (Energiagazdalkodasi Tudomanyos Egyesulet) Budapest, Hungary Vol. 11, no.11/12, Nov./Dec. 1958

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FODOR, G. (Budapest, XI., Budafoki ut 6-8)

The interpretation of characteristics of fundamental equations of the electromagnetic field. Periodica polytechn electr 3 no.3: (EEAI 10:1)

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(Electromagnetic fields) (Equations)

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Uncl.

20012, GY.

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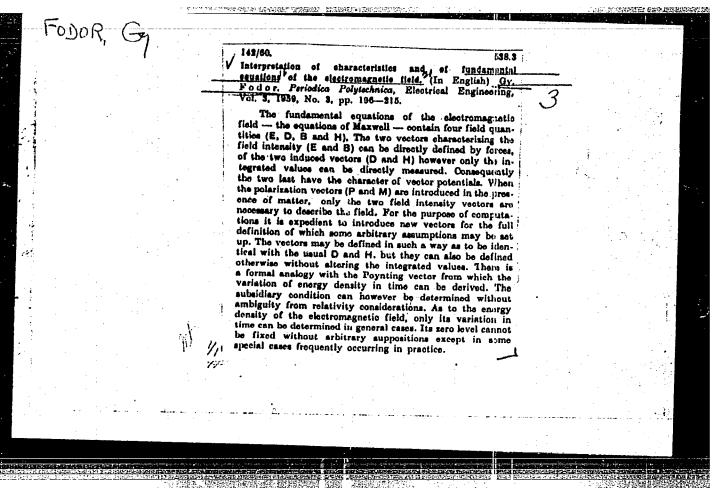
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FODOR, GY.

"A technical nuclear encyclopedia." p. 182.

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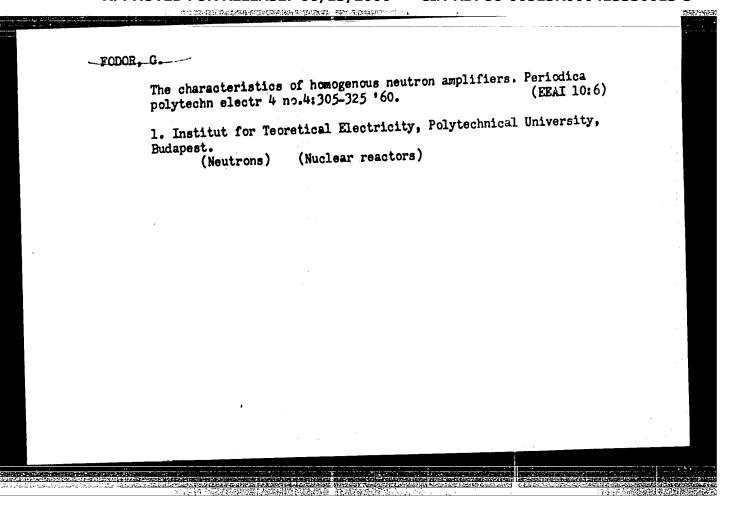
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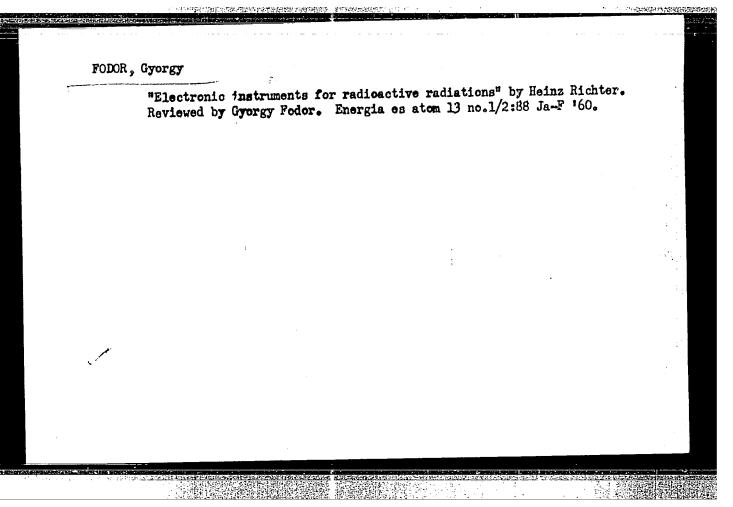
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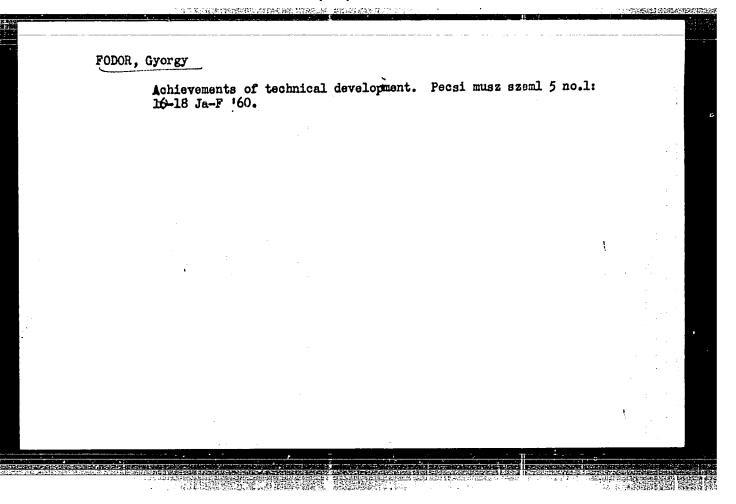




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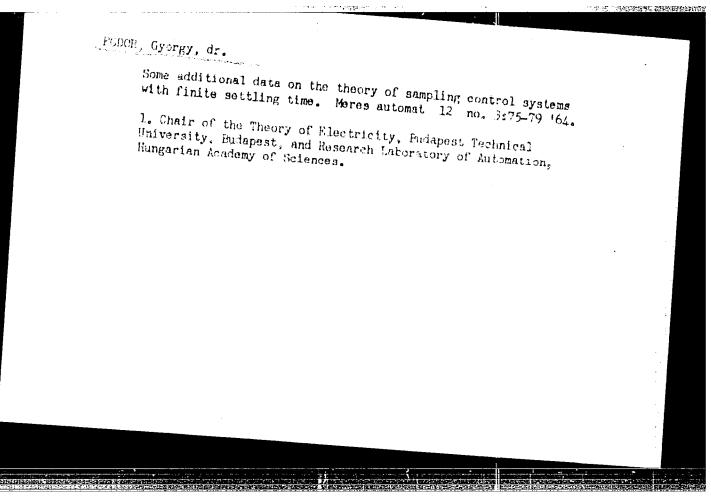
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